Study of Paramagnetic Chromocenes by Solid-State NMR Spectroscopy

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Summary: The application of ${}^{1}H$, ${}^{2}H$, and ${}^{13}C$ solid-state NMR spectroscopy to polycrystalline paramagnetic sandwich compounds was tested for chromocene ($Cp_{2}Cr$), deuteriated chromocene ($Cp_{2}Cr$ - d_{10}), and decamethylchromocene ($Cp *_{2}Cr$).

The bulk properties of paramagnetic compounds have been investigated thoroughly for a long time, and increasing effort is being observed particularly in molecular magnetism.¹ The commonly employed magnetic measurements integrate all over the sample, and individual information on different species being present is difficult to obtain. This kind of information is highly desirable, however, when paramagnetic catalysts are studied² and when molecule-based magnetic materials³ are designed from various building blocks. Our current organometallic approaches to both topics prompted us to further explore solid-state NMR spectroscopy and to select chromocenes (having two unpaired electrons) as prototypes.

Pioneering work on magic angle spinning (MAS) NMR of paramagnetic transition metal compounds has been carried out by several groups⁴ directed to both theory and application. But only very few studies of paramagnetic organometallic compounds are available.^{2d,5} Therefore, in addition to the background mentioned above, it seemed rewarding to initiate systematic work.

Polycrystalline chromocene Cp₂Cr, its deuteriated analogue Cp₂Cr- d_{10} , and decamethylchromocene Cp^{*}₂-

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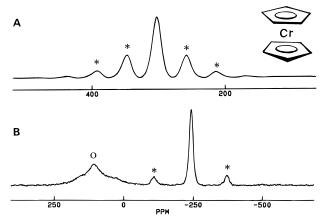


Figure 1. MAS NMR spectra of polycrystalline Cp₂Cr. For details of the measurements see text. Asterisks and the circle denote rotational sidebands and a probehead background signal, respectively. A: ¹H MAS NMR spectrum (300.1 MHz); rotational frequency 13 kHz. B: ¹³C MAS NMR spectrum (75.5 MHz); rotational frequency 10 kHz.

Cr⁶ were studied by ¹H, ²H, and ¹³C NMR spectroscopy as MAS and static samples. The ¹H and ¹³C MAS NMR signals of Cp₂Cr (Figure 1) cover ranges of 300 and 500 ppm, respectively; the shifts (and the half-widths) of the isotropic peaks are 315 ppm (4.9 kHz) and -258 ppm (1.2 kHz). The presence of only one ¹H and one ¹³C NMR signal suggests that corresponding nuclei of the two molecules of the unit cell⁷ are magnetically equivalent.⁸

The spectra of Figure 1 were measured with single pulse excitation in the high-power mode of a solid-state NMR spectrometer. Identical spectra were obtained in the low-power mode (with other parameters unchanged);⁹ the same was true of Cp_2Cr - d_{10} and Cp^*_2Cr . Hence, in general, spectra of this type can be also recorded by using conventional high-resolution NMR spectrometers equipped with a MAS probehead.

[†] Crystal structure analysis.

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⁽⁹⁾ Bruker MSL 300, 4 mm broadband double bearing MAS probehead, ca. 30 mg of polycrystalline material densely packed under N₂ and centered in ZnO₂ rotors by Teflon spacers, 90°-pulse widths for 10 μ s (²H), 6 μ s (¹³C), and 8 μ s (¹H), dead time delay 10 μ s, pulse spacing 100 ms. In the high-power mode the corresponding pulse widths were roughly three times smaller. Good S/N was obtained after 6000 (²H), 4000 (¹³C), and 100 (¹H) scans, i.e., after 10 min or less; line broadening factors of 20, 100, and 0 Hz, respectively were applied. Water and adamantane (low-frequency signal) were used as external standards.

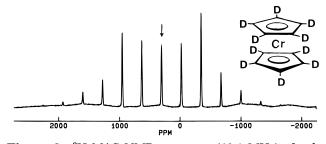


Figure 2. ²H MAS NMR spectrum (46.1 MHz) of polycrystalline Cp_2Cr - d_{10} . The isotropic line is denoted by an arrow; rotational frequency 13 kHz.

In order to assess the effect of the homo- and heteronuclear dipolar coupling on the line widths W of the individual signals within the spinning sideband manifold we investigated Cp_2Cr-d_{10} . Usually, when dipolar interactions are dominant, a dramatic reduction of Wis expected on passing from the protonated to the deuteriated compounds for both the ¹H^{4c,10} and ¹³C¹¹ MAS NMR spectra. However, only a minor decrease of $W(^{13}C)$ to 1.1 kHz was observed for Cp₂Cr- d_{10} at MAS frequencies of 10-15 kHz, while $W(^{1}H)$ of the residual protons did not change. This means that among the known different broadening factors^{4c,10,11} the interaction between the nuclei and the unpaired electrons determines the line widths W. Besides, MAS efficiently averages the internuclear dipolar interactions. Wshould also decrease strongly upon changing from ¹H to ²H NMR.^{11,12} In fact, in the ²H MAS NMR spectrum of Cp₂- $Cr-d_{10}$ (Figure 2) we find $W(^{2}H) = 0.7$ kHz for the peaks which constitute the expected Pake pattern typical of an I = 1 nucleus. This is seven times smaller than $W(^{1}H)$ of Cp₂Cr, but the effect is less pronounced than in solution.6c

The NMR spectra of the chromocenes depend strongly on the temperature and the rotational frequency $v_{\rm r}$. As commonly found for simple paramagnetic molecules, the magnitude of the signal shifts δ decreases on raising the temperature. By contrast, W remains essentially constant so that chromocenes may serve as solid-state NMR shift thermometers whose precision hardly depends on the temperature.4g Moreover, at a given bearing- and drive-gas temperature, the shift values $|\delta|$ decreased when the rotor frequency was increased. For example, raising v_r from 3 to 15 kHz changed δ ⁽¹³C) of Cp_2Cr from -258.4 to -231.3. This would be equivalent to increasing the sample temperature by 20 °C.¹³ The data obtained in variable MAS frequency studies could be fit to the expression $\delta = av_r^2 + b.^{14}$ This confirms previous work¹⁵ which attributed the effect to rotor

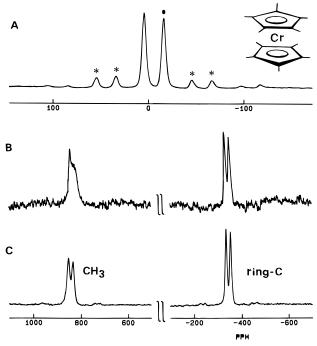


Figure 3. Solid-state NMR spectra of polycrystalline Cp*2-Cr. Asterisks denote rotational sidebands. A: ¹H MAS NMR spectrum (300.1 MHz); rotational frequency 15 kHz. B: ¹³C MAS NMR spectrum (75.5 MHz) recorded with irradiation⁹ at the low-frequency ¹H NMR resonance of spectrum A (denoted by the dot) during acquisition; rotational frequency 15 kHz. C: ¹³C MAS NMR spectrum (75.5 MHz) recorded with a simple one-pulse sequence; rotational frequency 8 kHz.

heating as a result of v_r^2 -dependent friction;^{15b} the relation $\delta \propto v_r$ was also reported.¹⁶

Further aspects of general interest were encountered when Cp*₂Cr was investigated. Both the ¹H and ¹³C MAS NMR spectra (Figure 3A,C) showed a splitting of the expected signals. In order to explain this result, which differs from that obtained for Cp₂Cr, we have redetermined¹⁷ the crystal structure. Cp^*_2Cr crystallizes in the space group $P2_12_12_1$. Two molecules A and B out of eight in the unit cell (Figure 4) are crystallographically and magnetically nonequivalent.

It is noteworthy that cross polarization was not efficient for chromocenes, probably owing to fast proton relaxation. Apart from this, simple high-power proton decoupling led to a striking broadening of ¹³C MAS NMR signals which proved to be selective. Thus, irrespective of the functional group, carbon atoms that bind protons were affected, whereas quarternary carbon atoms were not. In addition, the broadening could be restricted to one of the nonequivalent molecules in the unit cell; i.e., the broadening could be made siteselective. This is demonstrated in Figure 3: Irradiation at the low-frequency ¹H NMR signal of Cp*₂Cr (Figure 3A)¹⁸ led to a broadening of only one of the two ¹³C NMR signals (again that at low frequency) of the CH₃ groups

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⁽¹²⁾ Laukin, G.; Noack, F. Z. *Phys.* **1960**, *159*, 311. (13) As derived from temperature-dependent NMR studies of Cp₂-Cr in toluene- d_{s} : Cao, R.; Köhler, F. H. Unpublished data. (14) Cp₂Cr: $\delta^{(13C)} = 0.144\nu_r^2 - 263.9$ for $\nu_r \ge 6$ kHz, $\delta^{(1H)} = -0.111\nu_r^2 + 332.3$ for $\nu_r \ge 8$ kHz. Cp*₂Cr, ring-C atoms: $\delta^{(13C)} = 0.166\nu_r^2 - 364.9$ for $\nu_r \ge 2$ kHz. Cp*₂Cr, CH₃: $\delta^{(13C)} = -0.298\nu_r^2 + 870.4$ and $\delta^{(13C)} = -0.292\nu_r^2 + 851.2$ for ν_r ≥ 2 kHu ν_s^2 in kHz². ≥ 3 kHz; v_t^2 in kHz². v_r was incremented in 1 kHz steps up to 15 kHz. The correlation was good in all cases ($r^2 > 0.995$), the bearing- and drive-gas temperature was 298 K. Below the indicated MAS frequencies the signal shifts in the same order as given above were -258.4, 315.2, -345.2, -363.5, 867.9, and 847.5 ppm. Cp*₂Cr: $\delta(^{1}\text{H}) = 5.0$ and -16.Í.

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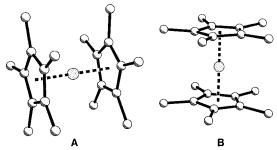


Figure 4. Plot of the crystallographically nonequivalent molecules A and B of Cp_2^*Cr . Selected averaged distances (Å) and angles (deg) for A/B: $C-CH_3$ 1.505(6)/1.502, C-C 1.415(6)/1.421(6), Cr-D 1.786(6)/1.779(6), D-Cr-D 179.8(2)/179.4(2). D is the center of the Cp* ligands. Bending range of the CH₃ groups out of the Cp planes: 0.5–4° (away from Cr). Average torsional angles (C–D–D–C) of the Cp* ligands of A/B: 17/–7 deg. The Cp* planes of A and those of B include an average angle of 89°.

(Figure 3B). Note that the signals of the ring carbon atoms did not change.

Among the origins of this effect, smearing out of the signals due to a temperature drift (especially on decoupling) can be ruled out, because the broadening is selective while the temperature behavior¹⁴ of the signal pairs is very similar. Rather the interference between the decoupling frequency and the rotation of the Cp ligands is expected to cause the signal broadening.¹⁹ For diamagnetic molecules such an interference was first demonstrated by Waugh,^{20a} and recent applications to the study of molecular motions were published by

Günther^{20b} and Griffin.^{20c} In the case of $Cp*_2Cr$ the interference should broaden the signals of CH_3 much more than that of the ring-C atoms, because the dipolar coupling is proportional to r^{-6} , r being the C···H distance. This was in fact observed.

In conclusion a promising entry into solid-state NMR studies of paramagnetic metallocenes was established. It includes equivalent and nonequivalent molecules in the unit cell, the correlation of the signal shift with the square of the MAS frequency, and chemical group-selective as well as site-selective ¹³C MAS NMR signal broadening by high-power proton decoupling. The latter phenomenon is expected to be generally applicable to the assignment of MAS NMR signals of paramagnetic compounds.

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Supporting Information Available: Tables of crystal data, atomic positional parameters, and bond distances, bond angles, dihedral angles between least-squares planes, and anisotropic displacement parameters and three figures of $Cp*_2Cr$ (12 pages). Ordering information is given on any current masthead page.

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⁽¹⁸⁾ A high MAS frequency was necessary in order to minimize the overlap of the proton signals due to rotational sidebands, and reduced decoupling power⁹ was applied.

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